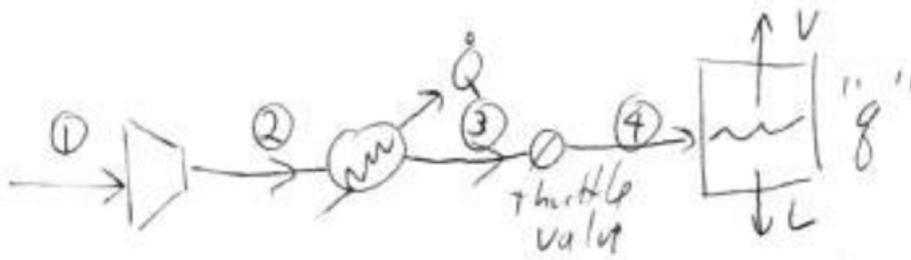


Quiz 6
Chemical Engineering Thermodynamics
February 16, 2017

Homework 4.30 involved a two-stage compressor with inter-stage cooling to produce liquid propane. The result of that analysis was $W_I = 150$ kJ/kg, $W_{II} = 250$ kJ/kg, $Q_I = 160$ kJ/kg, and $Q_{II} = 560$ kJ/kg with the final product having $q = 0.48$ (51% liquid). It was stated in class that a multistage compressor is more efficient compared to a single stage compressor. Redo this calculation for a single stage compressor using the following table for input parameters



	State	T, °F	P, Mpa	$\eta\theta$	q	H, kJ/kg	S, Btu/lb-F°
1	V	80	0.1		-		
2'	V		4.5		-		
2	V		4.5	0.80	-		
3	V	116			-		
4	V/L		0.1				

- Obtain the work for the compressor and compare it with the sum of the work for the two compressors given above.
- Obtain the cooling, Q , and compare it with the sum of the Q 's from the two stage compressor.
- Finally, obtain the fraction liquefied in the last stage and compare it with the fraction obtained from the two stage compressor.
- Is the dual stage compression more efficient? What is the downside to dual stage compression?

Quiz 6

	State	$T, ^\circ\text{F}$	P, MPa	η_0	g	H kJ/kg	S $\text{Btu/lb-}^\circ\text{F}$
1	V	80	0.1		-	980	1.47
2'	V	350	4.5		-	1200	1.47
2	V	390	4.5	0.8	-	1260	1.50
3	L	116	4.5	0	-	640	1.12
4	L/V	80	0.1		0.48	640	1.15

$$W_I' = 1200 \frac{\text{kJ}}{\text{kg}} - 980 \frac{\text{kJ}}{\text{kg}} = 220 \frac{\text{kJ}}{\text{kg}}$$

$$W_2 = \frac{220 \frac{\text{kJ}}{\text{kg}}}{0.8} = 275 \frac{\text{kJ}}{\text{kg}}$$

$$H_2 = 980 + 275 = 1255 \frac{\text{kJ}}{\text{kg}}$$

$$Q_I = 1260 \frac{\text{kJ}}{\text{kg}} - 640 \frac{\text{kJ}}{\text{kg}} = 620 \frac{\text{kJ}}{\text{kg}}$$

$$H_{0.1 \text{ MPa}}^V = 850 \frac{\text{kJ}}{\text{kg}}$$

$$H_{0.1 \text{ MPa}}^L = 430 \frac{\text{kJ}}{\text{kg}}$$

$$640 \frac{\text{kJ}}{\text{kg}} = 430 \frac{\text{kJ}}{\text{kg}} + g(850 \frac{\text{kJ}}{\text{kg}} - 430 \frac{\text{kJ}}{\text{kg}})$$

$$\frac{640 \frac{\text{kJ}}{\text{kg}} - 430 \frac{\text{kJ}}{\text{kg}}}{420 \frac{\text{kJ}}{\text{kg}}} = g$$

$$g = 0.48$$

g is the same between dual & single stage Compressor
52% Liquid

ΣW is $400 \frac{\text{kJ}}{\text{kg}}$ for 2 stage & $275 \frac{\text{kJ}}{\text{kg}}$ for single stage

ΣQ is $720 \frac{\text{kJ}}{\text{kg}}$ for 2 stage & $620 \frac{\text{kJ}}{\text{kg}}$ for single stage

E.11 PRESSURE-ENTHALPY DIAGRAM FOR PROPANE

(Source: NIST, Thermophysics Division, Boulder, CO, USA, used with permission.)

